## MARK SCHEME for the October/November 2014 series

## **0653 COMBINED SCIENCE**

0653/33

Paper 3 (Extended Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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Page 2		Mark Scheme Syllabus			
	-	Cambridge IGCSE – October/November 2014	0653	Paper 33	
1	(a) (i)	iron has reacted with oxygen in the air/water takes the place of oxy that has reacted with the iron ;	ygen	[1]	
	(ii)	iron has not reacted with helium/helium is unreactive ;		[1]	
	(b) (i)	same number of electrons ; same number of electron shells ; full electron shells/reference to complete outer shell ;		[max 2]	
	(ii)				
		2,8,1 configuration ;		[1]	
	(iii)	sodium atom has lost an electron ;		[1]	
	(iv)	(no reaction) sodium ions have electron configuration with full outer shell/sodiun do not gain or lose electrons ;	n ions	[1]	
		ne and use of noble gas ; perty related to use ;		[2]	
				[Total: 9]	
2	(a) (i)			[1]	
	(ii)	T ; T is the weight of canoe and man/description of downward force de gravity/the Earth;	ue to	[2]	

Pa	age 3	Mark Scheme	Syllabus	Paper
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	(iii)			
		peed		
		time		
		line drawn steepest at first ; smooth curve levelling off to horizontal ; horizontal section continuing ;		[3]
	<b>(b)</b> (tra	insfers to) thermal (heat)/movement of water/sound ;		[1]
		netic energy =) ½ mv²; ½ × 250 × 2 × 2 = 500 (J);		[2] [Total: 9]
3	(a) (i)	placenta correctly labelled ; cervix correctly labelled ;		[2]
	(ii)	glucose ; carbon dioxide ;		[2]
	(iii)	amniotic fluid ; cushions/protects/supports the fetus ;		[2]
	(b) (i)	amylase ✓ and x; protease ✓ and x;		[2]
	(ii)	digestion takes place in small intestine/enzymes are secreted here large intestine mainly absorbs water/enzymes not secreted here/ food already digested ;	e ;	[2]
	(de	stroys white blood cells ; stroys) T cells ; uces/weakens immunity ;		[max 2]
				[Total: 12]

P	age	4	Mark Scheme	Syllabus	Paper
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4	(a)		nplete circuit + switch ;		
		cor	rect parallel connection ;		[2]
	(b)	air/ attr mo	ter molecules move faster/has increased kinetic energy as they are 'owtte ; action forces between more water molecules are broken ; re water molecules have enough energy to escape (from water/hair flow removes escaped molecules so cannot return to hair/owtte ;		varm [max 3]
	(c)		molecules further apart as temperature rises ; ated) air becomes less dense (than surrounding air), so rises ;		[2]
	(d)	(i)	watt(s) ;		[1]
		(ii)	<i>I</i> = (P/V) = 1100/220 (= 5A) ;		[1]
		• •			
	(e)	(i)	short circuit (accept other reasonable ideas which might lead to fus e.g. too much current flowing in the circuit ;	e melting) ;	[1]
		(ii)	10A (no mark) 2A and 5A fuses would blow / 10A is the smallest fuse which will n 15A fuse gives less protection than 10A ;	ot flow ;	[2]
				I	Total: 12]
_					
5	(a)	(i)	geotropism ;		[1]
		(ii)	makes sure <u>roots</u> grow downwards/does not matter which way up (the roots will always grow downwards); to anchor plant;	the seed is	planted
			absorbs mineral ions/water ;		[max 2]
	(b)	reta	kins/the hormones inhibit slow down growth ; arded <u>cell elongation</u> where shaded/at bottom of the root ; is at top grow/expand normally/reference to differential growth ;		[max 2]

Page 5		5	Mark Scheme	Syllabus	Paper
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	(c)	(i)	no oxygen therefore slows or stops respiration ;		[1]
		(ii)	slows growth due to less/no energy being released ;		[1]
					[Total: 7]
6	(a)	calo wat	cium chloride ; er ;		[2]
	(b)	(i)	carbon dioxide lost from apparatus ; carbon dioxide gas has mass ;		[max 1]
		(ii)	rate decreases ; quickly at first then more slowly/stops at mass 203 g/after 6 minute (because) acid concentration decreases ; (because) surface area of calcium carbonate decreases ; reference to reduced collision frequency ;	9S ;	[max 3]
	(c)	(i)	203g;		[1]
		(ii)	particles have more (kinetic) energy/move faster at higher tempera collide more frequently ;	iture ;	
			increased chance of successful collision ;		[max 2]
					[Total: 9]
7	(a)	(i)	visible light ; radio waves (and) ultra-violet (both required for mark) ;		[2]
		(ii)	reflection ;		[1]
	(b)	(i)			

, (-)	gamma radiation	<b>X</b> ;		microwaves	
	Marana and li		 		[1]

(ii) X-rays and light will reach the Earth at the same time ;
all electromagnetic radiation travels at same speed (in vacuo);
[2]

[Total: 6]

Page 6		6	Mark Scheme	Syllabus	Paper		
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8	(a)	as the light intensity decreases the rate of photosynthesis decreases/ora ; not a linear/proportional relationship/numbers taken from graph to illustrate relationship ; [2]					
	(b)		ter rate with plant <b>P</b> (than plant <b>Q</b> ) or vice versa because it gets mor ter/plants/debris prevent some light from reaching plant <b>Q</b> ;	e light ;	[2]		
	(c)	(i)	causes surface plants/algae to grow faster ;		[1]		
		(ii)	reduces light to plant Q ; little or no photosynthesis ; (leading to) reduced growth of plant/plant dies ;		[max 2]		
					[Total: 7]		
9	(a)	(i)	aluminium/oxygen is an element because it/an element, consists type of atom ; aluminium oxide is a compound because it/a compound contains o elements bonded together ;		ms/ [max 1]		
		(ii)	bauxite is a mixture because it has a variable composition/can be aluminium oxide is a compound because it contains a fixed proport elements/can only be separated by chemical methods ;		[max 1]		
	(b)	Al₂ idea	O <sub>3</sub> ; a of balanced charges ;		[2]		
	(c)	aluminium <u>ions</u> migrate/move to/go to are attracted to the negative electrode/cathode ; electrons flow from cathode to each aluminium ion ;					
		3 e	lectrons/aluminium ions are discharged ;		[max 3]		
	(d)	carbon is less reactive than aluminium/below aluminium in the reactivity series/aluminium is more reactive than carbon/above carbon in the reactivity series/copper is less reactive than carbon;					
			bon will not react with/reduce/remove oxygen from aluminium oxide place aluminium ;	e/carbon wi	ll not [2]		
					[Total: 0]		

[Total: 9]